

REFORE COMPLETING FO

REPORT DOCUMENTATION PAGE AD-A096 54 200014-80-0-0305-2 TITLE 10 2 Sabarbas 5 TYPE OF REPORT & PERIOD COVERED Raman Spectrum of Pressure Compacted Technical Report 2 Fused Silica E PERFORMING ONG. REPORT HUMBER WANTHORES 6 CONTRACT OR GRANT NUVSERY G. F. Walrafen N00014-80-C-0305 € P. N. Krishman 10 PRODURAN FLEMENT, PROJECT, TASK AREA & WORK UNIT HUMMERS SESECTA DA BALA HOLLA SINADE OBINERO PER 6 Department of Chemistry Howard University NR-051-733 Rashington, D. C. 20059 11. CONTROLLING OFFICE NAME AND ADDRESS 12. REPORT DATE Jan. 21, 1981 Office of Raval Research 13. NUMBER OF PAGES Department of the Navy Arlington, Virginia 22217
Ts MONITORING AGENCY HAVE & ADDRESSIT different from Controlling Office) 15. SECURITY CLASS (of this isport) Unclassified

16. DISTRIBUTION STATEMENT (of this Report)

Approved for public release; reproduction is permitted for any purpose of the United States government distribution is unlimited.

17. DISTRIBUTION STATEMENT fof the obstract entered in Black 20, Il different lies: Report)

Distribution of this document is unlimited.

ID. SUPPLEMENTARY HOTES

Prepared and accepted for publication in the Journal of Chemical Physics.

19. 'EY WORDS (Centinue on reverse elde if necessary and identify by block number)

Raman Spectroscopy, fused silica

15%, DECLASSIFICATION/DOWNGRADING SCHEDULE

20. ARSTRACT (Continue on reverse side if necessary and identity by block number)

D Raman Spectra have been obtained from fused silica irreversably Slight spectral sharpening was observed compacted at 90 kbar, similar to that seen in Raman studies of neutron densified fused silica.

DD 1 JAN 71 14/3

EDITION OF I NOV 65 IS OBSOLETE 578 0102-011-6601 :

SECURITY CLASSICATED OF THIS PAGE Globan Fran Colored

## OFFICE OF NAVAL RESEARCH

N00014-80-C-03054, NFF-11. 77

Task No. NR-051-733

7 Technical Report No. 2

Raman Spectrum of Pressure Compacted Fused Silica,

by
G. E./Walrafen

P. N./Krishnan

1- 71-

Howard University Department of Chemistry Washington, D. C. 20059

Reproduction in whole or in part is permitted for any purpose of the United States Government

This document has been approved for public release and sale; its distribution is unlimited

Acces	sion For				
NTIS	GRA&I	X			
DTIC	TAB				
Unannounced					
Justi	.fication				
Ву	·				
Distr	ibution/				
Avai	lability Co	odes			
-	Avail and/	or I			
Dist	Special				
^	1 1	i			
u	!!	Ì			
11	1 1	1			

"Raman Spectrum of Pressure Compacted

Fused Silica"

ъу

G. E. Walrafen
Department of Chemistry
Howard University
Washington, D. C. 20059

and

P. N. Krishnan
Department of Chemistry
Coppin State College
Baltimore, Maryland 21216

In 1953 Bridgman and Simon<sup>(1)</sup> produced an irreversibly compacted modification of fused silica. By applying pressures of 10 GPa (100 kpar) at room temperature, a vitreous material was produced that lacked an x-ray diffraction pattern and had a density of 2.6 g cm.<sup>-3</sup> (2) MacKenzie<sup>(3)</sup> later obtained infrared spectra of compacted fused silica, but only slight changes in the Si-O stretching frequency were uncovered. Raman spectra of compacted fused silica, however, do not appear to be available, despite the fact that such spectra may be obtained at 1 atm. Accordingly, the Raman spectrum from compacted fused silica was obtained in this work, Fig. 1.

The compacted glass studied here was prepared by W. A. Rocco in R. H. Wentorf's laboratory (4) by subjecting ordinary fused silica to a pressure of 9.0 GPa for 1 h at 23°C. Densities of the uncompacted glass, 2.22 g cm, -3 and of the compacted disk, 2.40 g cm, -3 cut from the parent rod, were determined using buoyancy methods. The densified glass disk was about 1 mm thick and 6 mm in diameter. It had an irregular layered appearance, but it was sufficiently stable to be mounted, and it readily transmitted 514.5 nm laser light.

Raman spectra were obtained from the compacted disk and the uncompacted rod under identical conditions, <u>e.g.</u>, laser power, amplifier gain, slit width  $(5 \text{ cm}^{-1})$ , sample positioning, and polarization,  $z(x \ \frac{x}{y})y$ , Fig. 1, etc. The argon ion laser beam was transmitted radially through the disk or rod, and the Raman radiation was collected along the rod axis.

The irregular layered nature of the sample produced severe laser light scattering, however, which greatly increased the stray light in the Instruments S. A. HG2S double monochromator. An unavoidably large baseline curvature resulted, upper spectrum, Fig. 1.

The Raman spectrum of the compacted glass disk (compacted) is compared to that from the uncompacted parent rod (normal) in Fig. 1. The gross spectral features of the two samples are very similar -- no significant frequency changes of the Si-O stretching bands at 1060 and 1200 cm<sup>-1</sup> are apparent, cf., Ref. (3). However, careful analysis of the low-frequency Raman region including comparisons employing baselines A, B, and C (upper spectrum), (5) indicates that: (1) the intense Raman feature nominally at 440 cm<sup>-1</sup> narrowed with compaction by about 18 cm, -1 from a FWHH value of 278  $cm^{-1}$  (normal), to 260  $cm^{-1}$  (compacted), (2) a general loss of intensity primarily to the low-frequency side of the 440 cm<sup>-1</sup> peak occurred, (5) (3) the weak shoulder on the 440 cm<sup>-1</sup> peak at about  $350-375 \text{ cm}^{-1}$  weakened, (5) and (4) the 440 cm<sup>-1</sup> peak shifted upward in frequency from  $432 \pm 2 \text{ cm}^{-1}$  (normal), to  $441 \pm 2 \text{ cm}^{-1}$ (compacted), a change of 5 - 13 cm. $^{-1}$  (It would also appear that a weak broad feature at 900 cm<sup>-1</sup> occurring as a foot on the 800 cm<sup>-1</sup> Raman peak from normal fused silica, is more appropriately described as a weak separate maximum in all of the present compacted spectra.)

Stolen et al. (7) and Bates et al. (8) observed a narrowing and loss of low-frequency intensity below 440 cm<sup>-1</sup> in the Raman spectrum of neutron compacted fused silica, see Ref. (7) Fig. 1. An upward

shift of the 440 cm<sup>-1</sup> peak position, (7,8) and a strong intensification of the 603-609 cm<sup>-1</sup> defect peak were evident. (7,8) Also, a slight change at 900 cm<sup>-1</sup> similar to that observed here was indicated, (8) and a 10 cm<sup>-1</sup> increase in the 800 cm<sup>-1</sup> peak position was related to an angular decrease of 4° in the mean Si-0-Si bridging angle. (7)

Raman observations of Refs. (7,8) are similar to ours, except that no obvious intensity changes occurred for either of the defect peaks at 604 and 490 cm.<sup>-1</sup> (9) Raman spectral effects of pressure compaction are phenomenologically similar (although weaker) to those of neutron compaction for some specific features of the main glass network, but dissimilar for defects. Broken bonds are known to exist in neutron compacted silica, (10) however, and these may indirectly increase the defect concentration and thus the 604 cm<sup>-1</sup> intensity. But broken bond and/or defect concentration changes do not seem to be compatible with the present results. Further, we see virtually no change in the 800 cm<sup>-1</sup> peak position, although some slight shape changes may exist between 750-950 cm.<sup>-1</sup>

In conclusion, pressure compaction produces some changes in the network structure of vitreous silica, but no sizeable changes in the defect structures. Defect concentrations, and hence defect intensities at 604 and 490 cm<sup>-1</sup> are increased by increases in the fictive temperature, (9) but apparently not by irreversible compaction. We also are reluctant to infer sizeable changes in the mean Si-O-Si bridging angle, or the mean Si-O bond length, although changes in the SiO<sub>4</sub> structures, or in the mean torsional angle between SiO<sub>6</sub> tetrahedra, are not ruled out.

This work was supported by an ONR laser-chemistry contract.

Raman instruments used were purchased through a NSF grant,

CHE77-09888. Helpful discussions with A. G. Revesz are greatly appreciated.

#### REFERENCES

- 1. P. W. Bridgman and I. Simon, J. Appl. Phys. 24, 405(1953).
- 2. Density of fused silica, about 2.2 g cm; $^{-3}$  quartz, about 2.65 g cm. $^{-3}$
- 3. J. D. MacKenzie, J. Amer. Ceram. Soc. 46, 461(1963).
- 4. General Electric Research Laboratory, Schnectedy, NY.
- 5. To compare the two spectra of Fig. 1 it was necessary to normalize them using amplitudes of three peaks between 440 and 604 cm<sup>-1</sup> to determine if differences occurred elsewhere. From peak height ratios at 604, 490, and 440 cm<sup>-1</sup> above baselines A and D, Fig. 1. and from ratios of vertical height differences between the 604 and 490 cm<sup>-1</sup> peaks, the scattering level of the uncompacted spectrum was determined to be 1.04 times greater than that of the compacted spectrum. The uncompacted Raman intensities from 0-1400 cm<sup>-1</sup> were then divided by 1.04 and subtracted from the corresponding compacted spectral intensities. This procedure yielded baseline C, which represents the baseline that the compacted sample would have, if no spectral effects had occurred. However, baseline C falls below baselines B and A at low frequencies, thus definitely indicating spectral changes. Baseline B, the straight sloping baseline, constitutes the low baseline limit at low Raman frequencies. Baseline B resulted from linear extrapolation of a small fluorescence contribution at the larger Raman frequencies. (6) Baseline A, above baseline B, is a realistic estimate. The difference between baselines A and C is a measure of the spectral changes, conclusions (2) and (3). Twice this difference is shown in Fig. 1 between 100-500 cm. -1

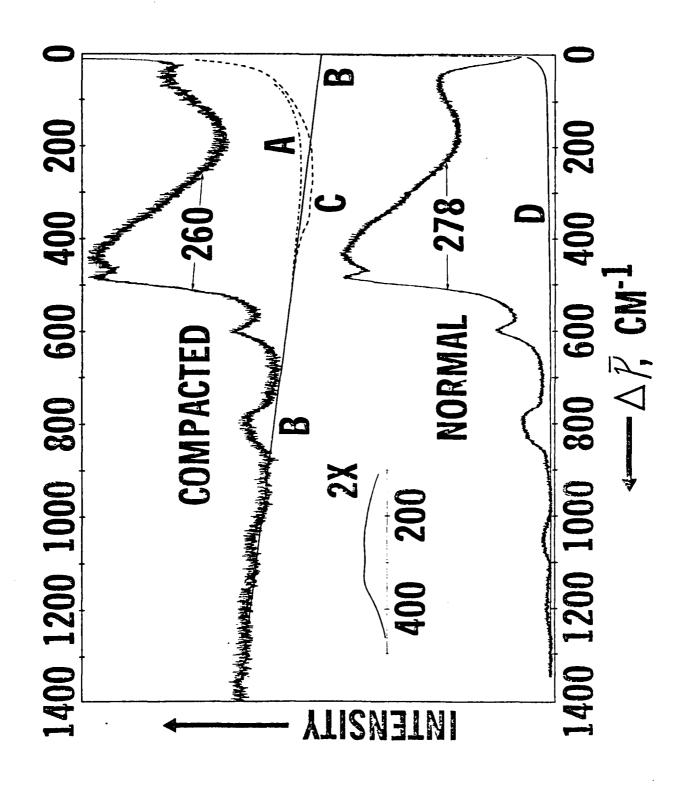
#### REFERENCES CONT'D.

- 6. The fluorescence observed here, unlike the strong fluorescence associated with neutron compacted fused silica, (7) is thought to be related to minor impurities in the compacted sample.
- 7. R. H. Stolen, J. T. Krause, and C. R. Kurkjian, Disc. Faraday Soc. No. 50, 103(1970).
- J. B. Bates, R. W. Hendricks and L. B. Shaffer, J. Chem. Phys.
   4163(1974).
- 9. R. H. Stolen and G. E. Walrafen, J. Chem. Phys. <u>64</u>, 2623(1976).
- 10. R. A. Weeks, Seventh Conf. on Glass, Brussels, 1965. R. A. Weeks and E. Sonder, "Paramagnetic Resonance. Vol. II."

  Academic Press, New York, 1963, p. 869.
- 11. Raman spectral effects of reversible compaction of fused silica are under investigation by M. Hokmabadi (doctoral dissertation), in this laboratory.

### CAPTION

and of normal fused silica (lower). A and D are realistic baseline estimates. B is a low baseline limit at low frequencies. Baseline C was obtained by subtracting the uncompacted intensity/1.04, from the compacted intensity. For clarity, the differences between baselines A and C were multiplied by 2, inset to left. This difference is a spectral measure of some effects of compaction. Values on arrows are full width at half height (FWHH) in cm. -1 Arrows are approximately parallel to baselines A and D.



# TECHNICAL REPORT DISTRIBUTION LIST, GEN

	No. Copies		No. Copies
Office of Naval Research		U.S. Frmy Research Office	
Attn: Code 472		Attn: CRD-AA-IP	
800 North Quincy Street		P.O Dx 1211	
Arlington, Virginia 22217	2	Research Triangle Park, N.C. 27709	1
ONR Branch Office		Naval Ocean Systems Center	
Attn: Dr. George Sandoz		Attn: Mr. Joe McCartney	
536 S. Clark Street		San Diego, California 92152	1
Chicago, Illinois 60605	1	Naval Weapons Center	
ONR Area Office		Attn: Dr. A. B. Amster,	
Attn: Scientific Dept.		Chemistry Division	
715 Broadway		China Lake, California 93555	1
New York, New York 10003	1	, , , , , , , , , , , , , , , , , , , ,	_
,	_	Naval Civil Engineering Laboratory	
ONR Western Regional Office		Attn: Dr. R. W. Drisko	
1030 East Green Street		Port Hueneme, California 93401	1
Pasadena, California 91106	1		
		Department of Physics & Chemistry	
ONE Eastern/Central Regional Office		Naval Postgraduate School	
Attn: Dr. L. H. Peebles		Monterey, California 93940	1
Building 114, Section D			
666 Summer Street		Dr. A. L. Slaikosky	
Boston, Massachusetts 02210	1	Scientific Advisor	
Director, Naval Research Laboratory		Commandant of the Marine Corps (Code RD-1)	
Attn: Code 6100		Washington, D.C. 20380	1
Washington, D.C. 20390	1	2000	_
2002	-	Office of Naval Research	
The Assistant Secretary		Attn: Dr. Richard S. Miller	
of the Navy (RE&S)		800 N. Quincy Street	
Department of the Navy		Arlington, Virginia 22217	1
Room 42736, Pentagon			
Washington, D.C. 20350	1	Naval Ship Research and Development Center	
Commander, Naval Air Systems Command		Attn: Dr. G. Bosmajian, Applied	
Attn: Code 310C (H. Rosenwasser)		Chemistry Division	
Department of the Navy		Annapolis, Maryland 21401	1
Washington, D.C. 20360	1		
		Naval Ocean Systems Center	
Defense Technical Information Center	•	Attn: Dr. S. Yamamoto, Marine	
Building 5, Cameron Station		Sciences Division	•
Alexandria, Virginia 22314	12	San Diego, California 91232	1
Dr. Fred Saalfeld		Mr. John Boyle	
Chemistry Division, Code 6100		Materials Branch	
Naval Research Laboratory		Naval Ship Engineering Center	
Washington, D.C. 20375	1	Philadelphia, Pennsylvania 19112	1

DTNSRDC Code 2803

Annapolis, Maryland 21402

# TECHNICAL REPORT DISTRIBUTION LIST, GEN

No.

1

	Copies
Dr. Rudolph J. Marcus	
Office of Naval Research	
Scientific Liaison Group	
American Embassy	
APO San Francisco 96503	1
Mr. James Kelley	

2

### TECHNICAL REPORT DISTRIBUTION LIST, 051A

	No. Copies		No. Copies
Dr. M. A. El-Sayed		Dr. M. Rauhut	
Department of Chemistry		Chemical Research Division	
University of California,		American Cyanamid Company	
Los Angeles		Bound Brook, New Jersey 08805	1
Los Angeles, California 90024	1		
		Dr. J. I. Zink	
Dr. E. R. Bernstein		Department of Chemistry	
Department of Chemistry		University of California,	
Colorado State University		Los Angeles	
Fort Collins, Colorado 80521	1	Los Angeles, California 90024	1
Dr. C. A. Heller		Dr. D. Haarer	
Naval Weapons Center		IBM	
Code 6059		San Jose Research Center	
China Lake, California 93555	1	5600 Cottle Road	
, , , , , , , , , , , , , , , , , , , ,		San Jose, California 95143	1
Dr. J. R. MacDonald		•	
Chemistry Division		Dr. John Cooper	
Naval Research Laboratory		Code 6130	
Code 6110		Naval Research Laboratory	
Washington, D.C. 20375	1	Washington, D.C. 20375	1
Dr. G. B. Schuster		Dr. William M. Jackson	
Chemistry Department		Department of Chemistry	
University of Illinois		Howard University	
Urbana, Illinois 61801	1	Washington, DC 20059	1
Dr. A. Adamson		De George E. Walraffen	
Department of Chemistry		Department of Chemistry	
University of Southern		Howard University	
California		Washington, DC 20059	1
Los Angeles, California 90007	1		

Dr. M. S. Wrighton
Department of Chemistry
Massachusetts Institute of
Technology
Cambridge, Massachusetts 02139

